- 1. Diamagnetic susceptibility of atomic hydrogen. The wave function of the hydrogen atom in its ground state (1s) is $\psi = (\pi a_0^3)^{-1/2} \exp(-r/a_0)$, where $a_0 = \hbar^2/me^2 = 0.529 \times 10^{-8}$ cm. The charge density is $\rho(x, y, z) = -e|\psi|^2$, according to the statistical interpretation of the wave function. Show that for this state $\langle r^2 \rangle = 3a_0^2$, and calculate the molar diamagnetic susceptibility of atomic hydrogen $(-2.36 \times 10^{-6} \text{ cm}^3/\text{mole})$.
- 2. Hund rules. Apply the Hund rules to find the ground state (the basic level in the notation of Table 1) of (a) Eu⁺⁺, in the configuration 4f⁷ 5s²p⁶; (b) Yb³⁺; (c) Tb³⁺. The results for (b) and (c) are in Table 1, but you should give the separate steps in applying the rules.
- 3. Triplet excited states. Some organic molecules have a triplet (S = 1) excited state at an energy k_BΔ above a singlet (S = 0) ground state. (a) Find an expression for the magnetic moment ⟨μ⟩ in a field B. (b) Show that the susceptibility for T ≥ Δ is approximately independent of Δ. (c) With the help of a diagram of energy levels versus field and a rough sketch of entropy versus field, explain how this system might be cooled by isentropic magnetization (not demagnetization).
- 8. Paramagnetism of S=1 system. (a) Find the magnetization as a function of magnetic field and temperature for a system of spins with S=1, moment μ , and concentration n. (b) Show that in the limit $\mu B \ll kT$ the result is $M \cong (2n\mu^2/3kT)B$.